

## New Cytotoxic Xenia Diterpenoids from the Formosan Soft Coral *Xenia umbellata*

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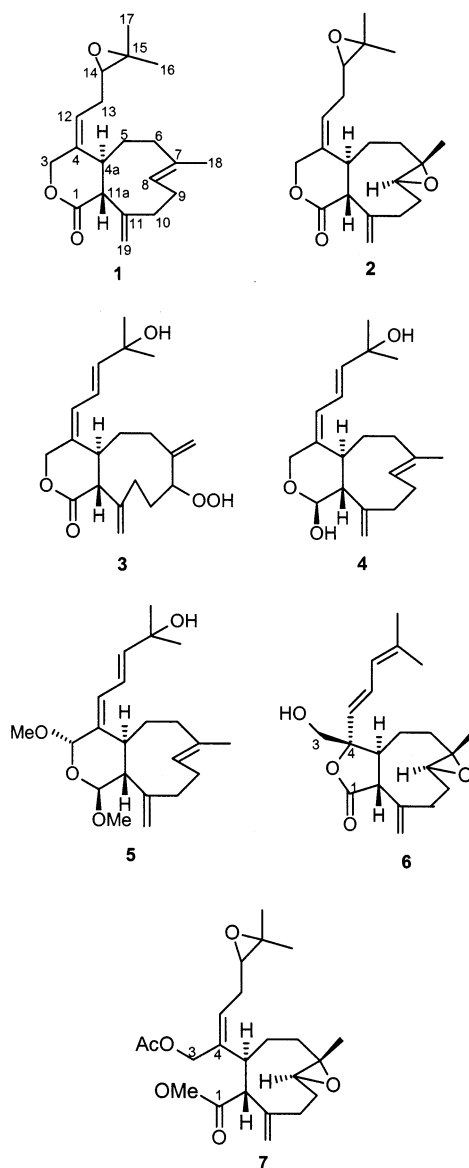
Received June 14, 2002

Seven new cytotoxic xenicane-type diterpenoids, 9-deoxyxeniloide-E (**1**), 9-deoxy-7,8-epoxyxeniloide-E (**2**), xeniolide-G (**3**), 9-deoxyxenialactol-C (**4**), xenibecin (**5**), xeniolide-H (**6**), and xenitacin (**7**), were isolated from the methylene chloride solubles of the Formosan soft coral *Xenia umbellata*. The structures were elucidated by 1D and 2D NMR spectral analysis, and their cytotoxicity against selected cancer cells was measured in vitro.

Soft corals belonging to the genus *Xenia* (order Alcyonacea, family Xeniidae) have proved to be a rich sources of terpenoids and have afforded several types of bioactive diterpenoids.<sup>1</sup> As part of our search for bioactive substances from marine organisms, the Formosan soft coral *Xenia umbellata* Lamarck was studied because the CH<sub>2</sub>Cl<sub>2</sub> extracts showed significant cytotoxicity to A549 (human lung adenocarcinoma), HT-29 (human colon adenocarcinoma), and P-388 (mouse lymphocytic leukemia) cell cultures as determined by standard procedures.<sup>2,3</sup> Bioassay-guided fractionations resulted in the isolation of seven new cytotoxic xenicane-type diterpenoids, 9-deoxyxeniloide-E (**1**), 9-deoxy-7,8-epoxyxeniloide-E (**2**), xeniolide-G (**3**), 9-deoxyxenialactol-C (**4**), xenibecin (**5**), xeniolide-H (**6**), and xenitacin (**7**), from *X. umbellata*.

### Results and Discussion

The molecular formula of **1** was established as C<sub>20</sub>H<sub>28</sub>O<sub>3</sub> by HREIMS. This formula, indicating seven degree of unsaturation, was fully supported by <sup>13</sup>C NMR and DEPT spectral data. Subsequent analysis of 2D NMR correlation data, focusing in particular on the interpretation of COSY, HMQC, and HMBC experiments, allowed the structure of **1** to be defined as 9-deoxyxeniolide-E. The <sup>13</sup>C NMR spectrum of **1** showed a carbonyl carbon at  $\delta$  173.3 and six additional olefinic carbons (three quaternary, two tertiary, one secondary), which accounted for 4 degrees of unsaturation. Hence, **1** was clearly a tricyclic diterpene. The <sup>1</sup>H NMR spectrum showed one methyl bearing a trisubstituted double bond [ $\delta$  1.64 (3H, br s); 5.39 (1H, dd,  $J$  = 12.0, 4.5 Hz)], one exomethylene [ $\delta$  4.91 (1H, s); 5.03 (1H, s)], two methyl groups on an oxygenated carbon [ $\delta$  1.35 (3H, s); 1.36 (3H, s)], two geminal lactone methylene protons [ $\delta$  4.44 (1H, d,  $J$  = 11.7 Hz); 4.84 (1H, d,  $J$  = 11.7 Hz)], a single allylic methine proton [ $\delta$  2.92 (1H, m)], an epoxymethine [ $\delta$  2.80 (1H, t,  $J$  = 6.3 Hz)], and a tertiary olefinic methine [ $\delta$  5.60 (1H, t,  $J$  = 7.5 Hz)]. Using proton-detected hetero-correlation NMR methods (HMQC and HMBC (Figure 1)), all protons were correlated with their respective carbons and the structural features of **1** were clearly assigned. The relative stereochemistry of 9-deoxyxeniolide-E was established by a NOESY experiment (Figure 2) and by compari-



sons of the relevant vicinal coupling constants with several xeniolides possessing identical partial structures.<sup>4–14</sup> The *E* configuration of the carbon–carbon double bond (C-4, 12) was determined by a NOESY correlation between H-3 and H-12. The *trans*-ring junction was assigned according to

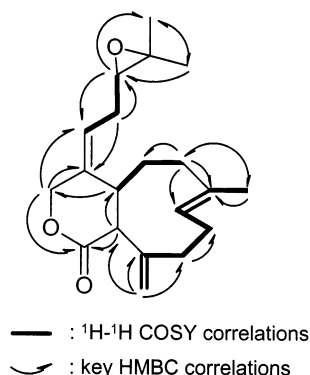
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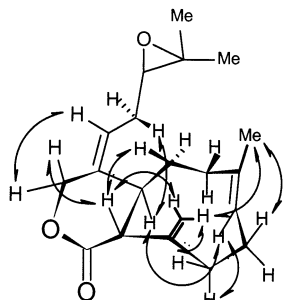
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**Figure 1.**  $^1\text{H}$ - $^1\text{H}$  COSY and key HMBC correlations of **1**.



**Figure 2.** Selected NOESY correlations of **1**.

NOESY correlations between H-4 $\alpha$  and H-8, between Me-18 and H-19, and between H-11 $\alpha$  and H-19 and H-11 $\alpha$  and H-3 $\alpha$ . NOESY correlation between H-4 $\alpha$  and H-11 $\alpha$  was not observed.

Compound **2** was analyzed for  $\text{C}_{20}\text{H}_{28}\text{O}_4$  by HREIMS and NMR spectral data. The NMR features of compound **2** were analogous to those of compound **1** with the exception that the resonances for the methyl-bearing *E*-trisubstituted olefin were replaced by those of a methyl-bearing *E*-trisubstituted epoxide ( $\delta_{\text{H}}$  1.21 s, 2.96 dd;  $\delta_{\text{C}}$  18.5 q, 58.6 s, 64.1 d). Cross-peaks in the  $^1\text{H}$ - $^1\text{H}$  COSY spectrum showed couplings between the epoxide methine proton at  $\delta$  2.96 (dd, H-8) and the methylene proton at  $\delta$  1.47 (dd, H-9 $\beta$ ) and 2.17 (m, H-9 $\alpha$ ). HMBC correlations between H-8 and C-6, C-7, C-9, C-10 and between Me-18 and C-6, C-7, C-8 positioned the methyl-bearing trisubstituted *E* epoxide at C-7, C-8, and C-18. The relative stereochemistry of **2** was established by a NOESY experiment (Supporting Information). NOESY correlations from H-11 $\alpha$  to H-3 $\beta$  ( $\delta$  4.87), H-19 ( $\delta$  5.19), and H-5 ( $\delta$  1.80) and from H-19 to Me-18 showed that these protons occurred on the same face of the ring system ( $\beta$ ). The coupling constant ( $J = 9.9$  Hz) between H-4 $\alpha$  and H-11 $\alpha$  suggested a *trans* ring junction, which implied that H-4 $\alpha$  was  $\alpha$ -oriented.<sup>5</sup> NOESY correlations from H-4 $\alpha$  to H-8 showed that these protons occurred on the same face of the ring system ( $\alpha$ ).

Compound **3** has the molecular formula  $\text{C}_{20}\text{H}_{28}\text{O}_5$ , as determined by HREIMS and NMR spectral data. The NMR spectra resembled those of **1**. However, a secondary hydroperoxy ( $\delta_{\text{H}}$  4.48 dd;  $\delta_{\text{C}}$  88.7 d)  $\alpha$  to an exocyclic methylene ( $\delta_{\text{H}}$  5.29 s, 5.37 s;  $\delta_{\text{C}}$  116.5 t, 149.4 s) in **3** replaced the *E*-trisubstituted double bond bearing a methyl group in **1**, and a side chain of xeniolide-B replaced the 14,15-epoxy terminus side chain in **1**.<sup>6</sup> COSY cross-peaks between H-8 and H-9 as well as HMBC correlations between H-8 and C-9, C-18; Me-18 and C-7, C-8; and H-9 and C-7, C-8, C-10 positioned the *exo*-methylene and secondary hydroperoxyl at C-7 and C-8, respectively. The coupling constant ( $J = 9.9$  Hz) between H-4 $\alpha$  and H-11 $\alpha$

suggested a *trans* ring junction. The relative stereochemistry of the secondary hydroperoxyl at C-9 was not determined due to the flexibility of the nine-membered ring.

The molecular formula  $\text{C}_{20}\text{H}_{30}\text{O}_3$  of compound **4** was revealed by HREIMS and NMR spectral data. The NMR features of compound **4** were analogous to those of xeniolactol-C with the exception that the resonance for the 9-hydroxyl methine was replaced by a methylene ( $\delta_{\text{H}}$   $\delta$  2.15, 2.49;  $\delta_{\text{C}}$  24.9).<sup>4</sup> HMBC correlations between H-18 and C-6, C-7, C-8; H-6 and C-5, C-7, C-18; and H-9 and C-7, C-8, C-11 confirmed the absence of a hydroxyl at C-9. In the NOESY experiment, NOEs between H-19 and H-1, H-4 $\alpha$ , and between H-4 $\alpha$  and H-8 allowed H-1, H-19, H-4 $\alpha$ , and H-8 to be assigned to the  $\alpha$ -face of the molecule. Further, NOEs between H-18 and H-9 $\beta$  and between H-11 $\alpha$  and H-3 $\beta$ /5 $\beta$  allowed H-18, H-9 $\beta$ , H-11 $\alpha$ , H-3 $\beta$ , and H-5 $\beta$  to be assigned to the  $\beta$ -face of the molecule. Additional NOESY correlations between H-4 $\alpha$  and H-13 and H-3 $\alpha$  and H-12 confirmed the *E* configuration of the carbon-carbon double bond at C-4, 12 (Figure 3).

The molecular formula of **5** was established as  $\text{C}_{22}\text{H}_{34}\text{O}_3$  by HREIMS and NMR spectral data. The NMR features of compound **5** were analogous to those of **4** with the exception that two methoxyl groups replaced the hydroxyl group at C-1 and the  $\alpha$ -methylene proton at C-3. HMBC correlations between H-1 and C-3, C-11, C-11 $\alpha$ ; H-3 and C-1, C-4, C-12, C-4 $\alpha$ ; OMe-1 and C-1; and OMe-3 and C-3 positioned the methoxyl groups at C-1 and C-3. The relative stereochemistry of **5** was established by a NOESY experiment. NOESY correlations from H-4 $\alpha$  to H-19/H-8 and from H-19 to H-1/9 $\alpha$  showed that these protons occurred on the same face of the ring system ( $\alpha$ ). NOESY correlations from H-11 $\alpha$  to H-3/5 $\beta$  and from Me-18 to H-9 $\beta$  showed that these protons occurred on the same face of the ring system ( $\beta$ ).

Compound **6** has the molecular formula  $\text{C}_{20}\text{H}_{28}\text{O}_4$ , as determined by HREIMS and NMR spectral data. The  $^1\text{H}$  NMR data of compound **6** were very close to those of florlide G isolated from the Japanese soft coral *Xenia florida* except chemical shifts for H-4 $\alpha$  (2.56 for **6**; 3.08 for florlide G) and H-3 (3.77 and 3.93 for **6**; 3.58 and 3.75 for florlide G).<sup>7</sup> HMBC correlations between H-3 and C-4/C-4 $\alpha$ /C-5 confirmed these assignments. The relative stereochemistry of **6** was established by a NOESY experiment (Figure 4). Correlations from H-4 $\alpha$  to H-12/H-10 $\alpha$ /H-6 $\alpha$  and from H-8 to H-6 $\alpha$ /H-10 $\alpha$  showed that these protons occurred on the same face of the ring system ( $\alpha$ ). Similarly, NOESY correlations from H-19 to H-11 $\alpha$ /Me-18 and from H-11 $\alpha$  to H-3 showed that these protons occurred on the same face of the ring system ( $\beta$ ).

The molecular formula  $\text{C}_{23}\text{H}_{34}\text{O}_6$  of compound **7** was revealed by HREIMS and NMR spectral data. The NMR data of compound **7** were analogous to those of florlide F isolated from the Japanese soft coral *Xenia florida* except that compound **7** had a C-14,15-epoxy terminus side chain.<sup>7</sup> HMBC correlations between H-14 and C-12, C-13, C-15, C-17 confirmed the position of the epoxy group.

The cytotoxicity of compounds **1**–**7** is shown in Table 3. Compound **7** exhibited cytotoxicity against P-388, HT-29, and A549 cells. Compound **3** showed potent cytotoxicity against P-388 cells maybe due to the hydroperoxide functionality.<sup>15</sup> Compounds **1** and **2** and **4**–**6** exhibited moderate cytotoxicity against P-388 cells.

## Experimental Section

**General Experimental Procedures.** Melting points were determined using a Yanagimoto micromelting point apparatus and are reported uncorrected. Optical rotations were determined on a JASCO DIP-181 polarimeter. UV spectra were

**Table 1.**  $^1\text{H}$  NMR Spectral Data<sup>a</sup> of **1**–**7** in  $\text{CDCl}_3$ 

	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>5</b>	<b>6</b>	<b>7</b>
<b>1</b>				4.61 d (8.4)	4.31 d (9.0)		
3 $\alpha$	4.44 d (11.7)	4.45 d (12.0)	4.44 d (12.0)	4.69 d (13.6)	5.25 s	3.77 d (12.6)	3.52 d (4.2)
3 $\beta$	4.84 d (11.7)	4.87 d (12.0)	4.88 d (12.0)	4.30 d (13.6)		3.93 d (12.6)	
4a	2.92 m	2.85 m	3.20 m	2.86 d (10.5)	2.87 d (10.4)	2.56 m	3.44 m
5 $\alpha$	1.69 m	1.92 m	2.13 m	1.53 m	1.40 m	2.06 m	1.73 m
5 $\beta$	1.71 m	1.80 m	2.56 m	1.98 m	1.88 m	1.90 m	1.55 m
6 $\alpha$	2.16 m	1.36 m	2.87 m	2.34 m	2.36 m	1.05 m	1.17 m
6 $\beta$		2.16 m	2.20 m	2.37 m	2.34 m	2.08 m	1.99 m
<b>8</b>	5.39 dd (12.0, 4.5)	2.96 dd (11.4, 2.7)	4.48 dd (5.6, 3.2)	5.43 t (10.2)	5.41 t (10.1)	2.87 dd (12.0, 3.6)	3.01 dd (11.4, 1.8)
9 $\alpha$	2.12 m	2.17 m	2.26 m	2.15 m	2.15 m	2.19 m	2.13 m
9 $\beta$	2.46 m	1.47 m	1.97 m	2.09 m	2.46 m	1.38 m	1.43 m
10 $\alpha$	2.59 m	1.85 m	2.58 m	2.22 m	2.30 m	1.88 m	2.13 m
10 $\beta$	2.17 m	2.14 m	2.33 m	2.26 m	2.34 m	2.58 m	2.34 m
11a	2.92 m	3.22 d (9.9)	3.32 d (9.9)	1.80 m	1.69 m	3.77 d (12.6)	3.44 m
<b>12</b>	5.60 t (7.5)	5.76 t (7.5)	6.08 d (11.4)	5.84 d (10.5)	6.25 d (9.6)	5.51 d (15.3)	5.52 dd (13.8, 7.2)
<b>13</b>	2.35 m	2.32 m	6.40 dd (15.3, 11.4)	6.45 dd (14.7, 10.5)	6.49 dd (15.3, 9.6)	6.50 dd (15.3, 12.0)	2.48 m
		2.12 m					2.32 m
<b>14</b>	2.80 t (6.3)	2.80 dd (7.5, 4.8)	5.95 d (15.3)	5.84 d (14.7)	5.95 d (15.3)	5.79 d (12.0)	2.75 dd (7.5, 6.0)
<b>16</b>	1.36 s	1.34 s	1.38 s	1.36 s	1.34 s	1.78 br s	1.31 s
<b>17</b>	1.35 s	1.29 s	1.37 s	1.36 s	1.34 s	1.80 br s	1.31 s
<b>18</b>	1.64 br s	1.21 s	5.29 s	1.77 br s	1.73 br s	1.19 s	1.14 s
			5.37 s				
<b>19</b>	4.91 s	5.16 s	5.10 s	4.72 s	4.66 s	5.22 s	5.08 s
	5.03 s	5.19 s	5.29 s	4.84 s	4.78 s	5.25 s	5.29 s
OMe-1					3.38 s		3.56 s
OMe-3					3.55 s		
OAc							2.04 s
OOH-8			8.20 br s				

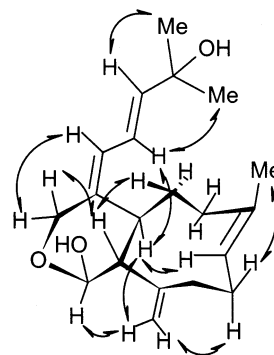
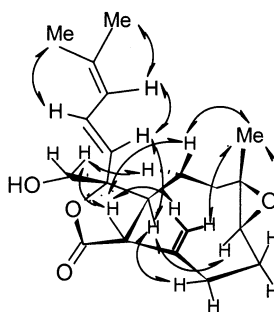
<sup>a</sup> Recorded in  $\text{CDCl}_3$  at 300 MHz.**Table 2.**  $^{13}\text{C}$  NMR Spectral Data<sup>a</sup> ( $\delta$ ) of **1**–**7** in  $\text{CDCl}_3$ 

	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>5</b>	<b>6</b>	<b>7</b>
<b>1</b>	173.3	172.6	173.0	100.0	104.5	176.6	172.5
<b>3</b>	71.7	71.4	71.6	69.8	99.4	65.6	65.2
<b>4</b>	138.6	138.8	135.1	140.1	139.5	88.2	138.2
4a	38.2	36.3	37.1	44.6	44.5	47.9	38.3
<b>5</b>	36.7	35.4	30.7	35.6	35.9	25.6	28.7
<b>6</b>	39.7	38.8	31.8	35.9	40.5	38.8	38.3
<b>7</b>	134.9	58.6	149.4	135.6	135.4	59.2	59.7
<b>8</b>	124.2	64.1	88.7	124.3	124.3	64.4	63.6
<b>9</b>	29.7	27.2	35.8	24.9	24.8	28.8	27.3
<b>10</b>	34.4	28.7	30.3	40.4	35.8	27.7	26.1
<b>11</b>	148.1	143.8	145.0	155.1	154.7	143.9	143.0
11a	57.0	57.4	50.9	57.6	56.2	59.2	60.6
<b>12</b>	125.9	126.0	128.4	121.9	124.5	126.3	126.6
<b>13</b>	28.4	28.9	120.8	121.1	121.0	127.7	28.1
<b>14</b>	62.8	62.6	145.2	142.0	144.0	123.9	63.2
<b>15</b>	58.6	59.4	71.1	71.0	71.0	138.8	58.8
<b>16</b>	24.8	24.7	29.9	29.9	29.9	26.1	24.8
<b>17</b>	18.9	18.9	30.0	30.0	29.9	18.6	18.9
<b>18</b>	17.8	18.5	116.5	16.7	16.7	18.2	18.8
<b>19</b>	116.9	119.8	118.2	110.7	110.3	121.6	121.2
OMe-1					57.0		51.7
OMe-3					55.3		
OAc							170.8
							21.1

<sup>a</sup> Recorded in  $\text{CDCl}_3$  at 75 MHz.

obtained on a Shimadzu UV-160A spectrophotometer, and IR spectra were recorded on a Hitachi 26-30 spectrophotometer. The NMR spectra were recorded on a Bruker Avance 300 NMR spectrometer at 300 MHz for  $^1\text{H}$  and 75 MHz for  $^{13}\text{C}$ , respectively, in  $\text{CDCl}_3$  using TMS as internal standard. EIMS spectra were obtained with a JEOL JMS-SX/SX 102A mass spectrometer at 70 eV. Si gel 60 (Merck, 230–400 mesh) was used for column chromatography; precoated Si gel plates (Merck, Kieselgel 60 F<sub>254</sub>, 0.25 mm) were used for TLC analysis.

**Animal Material.** The soft coral *Xenia umbellata* was collected at Green Island, off Taiwan, in April 2001, at a depth of 2–3 m and was stored for 1 month in a freezer until extraction. A voucher specimen, NSUGN-045, was deposited in the Department of Marine Resources, National Sun Yat-sen University, Taiwan.

**Figure 3.** Selected NOESY correlations of **4**.**Figure 4.** Selected NOESY correlations of **6**.

**Extraction and Isolation.** The bodies of the soft coral *X. umbellata* were freeze-dried to give 0.45 kg of a solid, which was extracted with  $\text{CH}_2\text{Cl}_2$  (3.0 L  $\times$  3). After removal of solvent in vacuo, the residue (30 g) was chromatographed over Si gel 60 using *n*-hexane and *n*-hexane–EtOAc mixtures of increasing polarity. Elution by *n*-hexane–EtOAc (6:1) afforded fractions containing compound **7**. Elution by *n*-hexane–EtOAc (4:1) afforded fractions containing compounds **1** and **5**. Elution by *n*-hexane–EtOAc (3:1) afforded fractions containing compounds **2** and **6**. Elution by *n*-hexane–EtOAc (2:1) afforded fractions containing compound **4**. Elution by EtOAc afforded fractions containing compound **3**. Compounds **1** and **5** were further purified by Si gel column chromatography, by eluting with *n*-hexane–acetone (9:1). Compounds **2** and **6** were further

**Table 3.** Cytotoxicity<sup>a</sup> of **1–7**

compound	cell line ED <sub>50</sub> (μg/mL)		
	A549	HT-29	P-388
<b>1</b>	11.2	21.1	2.87
<b>2</b>	11.6	7.77	3.35
<b>3</b>	4.77	8.31	0.04
<b>4</b>	4.85	12.9	3.45
<b>5</b>	13.4	12.5	3.96
<b>6</b>	18.8	5.33	3.66
<b>7</b>	3.26	1.12	1.09

<sup>a</sup> For significant activity of pure compounds, an ED<sub>50</sub> of 4.0 μg/mL is required.

purified by Si gel column chromatography by eluting with *n*-hexane–EtOAc (6:4).

**9-Deoxyxeniloide-E (1):** oil (23 mg); [α]<sub>D</sub><sup>25</sup> +16.8° (*c* 0.46, CHCl<sub>3</sub>); UV (MeOH) λ<sub>max</sub> nm (log ε) 206 (3.98); IR (KBr) ν<sub>max</sub> 1736, 1608, 1236 cm<sup>-1</sup>; <sup>1</sup>H NMR, see Table 1; <sup>13</sup>C NMR, see Table 2; EIMS *m/z* 316 [M]<sup>+</sup> (1), 300 (1), 272 (4), 43 (100); HREIMS *m/z* 316.2038 (calcd for C<sub>20</sub>H<sub>28</sub>O<sub>3</sub>, 316.2031).

**9-Deoxy-7,8-epoxyxeniloide-E (2):** oil (7 mg); [α]<sub>D</sub><sup>25</sup> +20.6° (*c* 0.23, CHCl<sub>3</sub>); UV (MeOH) λ<sub>max</sub> nm (log ε) 204 (3.76); IR (KBr) ν<sub>max</sub> 1734, 1610, 1240 cm<sup>-1</sup>; <sup>1</sup>H NMR, see Table 1; <sup>13</sup>C NMR, see Table 2; EIMS *m/z* 332 [M]<sup>+</sup> (1), 316 (1), 300 (1), 272 (3), 55 (100); HREIMS *m/z* 332.1975 (calcd for C<sub>20</sub>H<sub>28</sub>O<sub>4</sub>, 332.1980).

**Xeniolide-G (3):** oil (12 mg); [α]<sub>D</sub><sup>25</sup> +27.5° (*c* 0.40, CHCl<sub>3</sub>); UV (MeOH) λ<sub>max</sub> nm (log ε) 222 (3.88); IR (KBr) ν<sub>max</sub> 3360, 1732, 1621, 1230 cm<sup>-1</sup>; <sup>1</sup>H NMR, see Table 1; <sup>13</sup>C NMR, see Table 2; EIMS *m/z* 348 [M]<sup>+</sup> (1), 332 (3), 292 (5), 91 (100); HREIMS *m/z* 348.1922 (calcd for C<sub>20</sub>H<sub>28</sub>O<sub>5</sub>, 348.1929).

**9-Deoxyxeniaolactol-C (4):** amorphous solid (26 mg); [α]<sub>D</sub><sup>25</sup> -18.8° (*c* 0.52, CHCl<sub>3</sub>); UV (MeOH) λ<sub>max</sub> nm (log ε) 221 (3.79); IR (KBr) ν<sub>max</sub> 3305, 1616 cm<sup>-1</sup>; <sup>1</sup>H NMR, see Table 1; <sup>13</sup>C NMR, see Table 2; EIMS *m/z* 318 [M]<sup>+</sup> (1), 300 (2), 287 (2), 43 (100); HREIMS *m/z* 318.2195 (calcd for C<sub>20</sub>H<sub>30</sub>O<sub>3</sub>, 318.2187).

**Xenibecin (5):** oil (47 mg); [α]<sub>D</sub><sup>25</sup> -18.2° (*c* 0.24, CHCl<sub>3</sub>); UV (MeOH) λ<sub>max</sub> nm (log ε) 218 (3.96); IR (KBr) ν<sub>max</sub> 3420, 1610 cm<sup>-1</sup>; <sup>1</sup>H NMR, see Table 1; <sup>13</sup>C NMR, see Table 2; EIMS *m/z* 362 [M]<sup>+</sup> (1), 344 (1), 314 (3), 284 (4), 43 (100); HREIMS *m/z* 362.2443 (calcd for C<sub>22</sub>H<sub>34</sub>O<sub>4</sub>, 362.2448).

**Xeniolide-H (6):** oil (18 mg); [α]<sub>D</sub><sup>25</sup> +16.3° (*c* 0.46, CHCl<sub>3</sub>); IR (KBr) ν<sub>max</sub> 3500, 1742, 1608 cm<sup>-1</sup>; <sup>1</sup>H NMR, see Table 1; <sup>13</sup>C NMR, see Table 2; EIMS *m/z* 332 [M]<sup>+</sup> (1), 314 (2), 298 (3), 55 (100); HREIMS *m/z* 332.1988 (calcd for C<sub>20</sub>H<sub>28</sub>O<sub>4</sub>, 332.1980).

**Xenitacin (7):** oil (2 mg); [α]<sub>D</sub><sup>25</sup> +12.3° (*c* 0.12, CHCl<sub>3</sub>); UV (MeOH) λ<sub>max</sub> nm (log ε) 211 (4.02); IR (KBr) ν<sub>max</sub> 1740, 1730, 1600, 1240 cm<sup>-1</sup>; <sup>1</sup>H NMR, see Table 1; <sup>13</sup>C NMR, see Table 2; EIMS *m/z* 406 [M]<sup>+</sup> (2), 388 (3), 374 (8), 346 (5), 71 (100); HEIMS *m/z* 406.2337 (calcd for C<sub>23</sub>H<sub>34</sub>O<sub>6</sub>, 406.2346).

**Cytotoxicity Testing.** P-388 cells were kindly supplied by J. M. Pezzuto, Department of Medicinal Chemistry and Pharmacognosy, University of Illinois at Chicago; A549 and HT-29 were purchased from the American Type Culture Collection. Cytotoxic assays were carried out according to previously described procedures.<sup>3</sup>

**Acknowledgment.** We thank J. M. Pezzuto, Department of Medicinal Chemistry and Pharmacognosy, College of Pharmacy, University of Illinois at Chicago, for providing the P-388 cell line. This work was supported by grants from the National Science Council of Taiwan awarded to C.-Y.D.

**Supporting Information Available:** Tables of HMBC correlations and NOESY correlations of **1–7**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

## References and Notes

- Faulkner, D. J. *Nat. Prod. Rep.* **2001**, *18*, 1–49, and references therein.
- Geran, R. I.; Greenberg, N. H.; MacDonald, M. M.; Schumacher, A. M.; Abbott, B. J. *Cancer Chemother. Rep.* **1972**, *3*, 1–91.
- Hou, R.-S.; Duh, C.-Y.; Chiang, M. Y.; Lin, C.-N. *J. Nat. Prod.* **1995**, *58*, 1126–1130.
- Groweiss, A.; Kashman, Y. *Tetrahedron* **1983**, *39*, 3385–3396.
- Iwagawa, T.; Kawasaki, J.; Hase, T. *J. Nat. Prod.* **1998**, *61*, 1513–1515.
- Kashman, Y.; Groweiss, J. *Org. Chem.* **1980**, *45*, 3814–3824.
- Iwagawa, T.; Nakamura, K.; Hirose, T.; Okamura, H.; Nakatani, M. *J. Nat. Prod.* **2000**, *63*, 468–472.
- Miyaoka, H.; Nakano, M.; Iguchi, K.; Yamada, Y. *Tetrahedron* **1999**, *55*, 12977–12982.
- Miyaoka, H.; Mitome, H.; Nakano, M.; Yamada, Y. *Tetrahedron* **2000**, *56*, 7737–7740.
- Rho, J.; Lee, H.; Seo, Y.; Cho, K. W.; Shin, J. *J. Nat. Prod.* **2000**, *63*, 254–257.
- Vanderah, D. J.; Steudler, P. A.; Ciereszho, L. S.; Schimits, F. J.; Ekstrand, J. D.; van der Heim, D. *J. Am. Chem. Soc.* **1977**, *99*, 5780–5784.
- Iwagawa, T.; Amano, Y.; Hase, T.; Shiro, M. *Tetrahedron* **1995**, *51*, 11111–11118.
- Iwagawa, T.; Amano, Y.; Nakatani, M.; Hase, T. *Bull. Chem. Soc. Jpn.* **1996**, *69*, 1309–1312.
- Anta C.; González, N.; Santafé, G.; Rodríguez, J.; Jiménez, C. *J. Nat. Prod.* **2002**, *65*, 766–768.
- Duh, C.-Y.; Chia, M.-C.; Wang, S.-K.; Chen, H.-J.; El-Gamal, A. A. H.; Dai, C.-F. *J. Nat. Prod.* **2001**, *64*, 1028–1031.

NP020268Z